Oxidative Phosphorylation

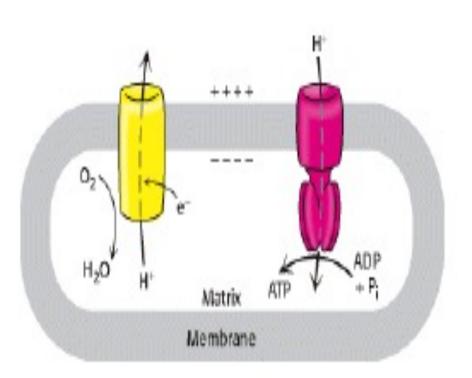
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INTRODUCTION

- Oxidative phosphorylation is the process in which ATP is formed as a result of the transfer of electrons from NADH or FADH2 to O2 by a series of electron carriers.
- This process, which takes place in mitochondria, is the major source of ATP in aerobic organisms
- Oxidative phosphorylation is conceptually simple and mechanistically complex

- Oxidative phosphorylation is the culmination of a series of energy transformations that are called cellular respiration
- First, carbon fuels are oxidized in the citric acid cycle to yield electrons with high transfer potential → this electron-motive force is converted into a proton-motive force and → the proton-motive force is converted into phosphoryl transfer potential.
- The conversion of electron-motive force → proton-motive force is carried out by three electron-driven proton pumps NADH-Q oxidoreductase, Q-cytochrome c oxidoreductase, and cytochrome c oxidase
- The final phase of oxidative phosphorylation is carried out by *ATP synthase,* an ATP-synthesizing assembly that is driven by the flow of protons back into the mitochondrial matrix.

Essence of Oxidative Phosphorylation



 Oxidation and ATP synthesis are coupled by transmembrane proton fluxes.

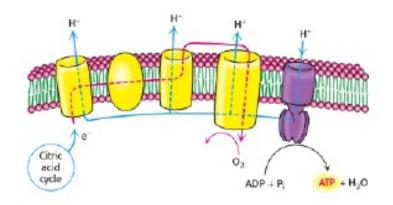
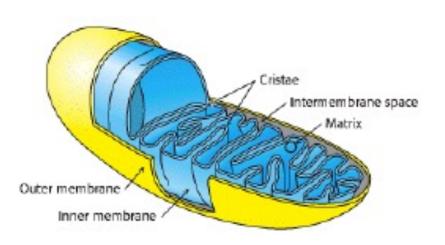
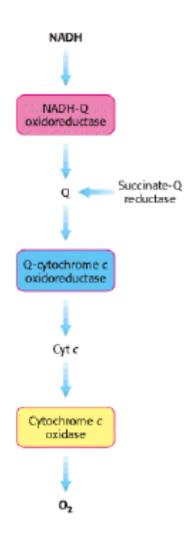


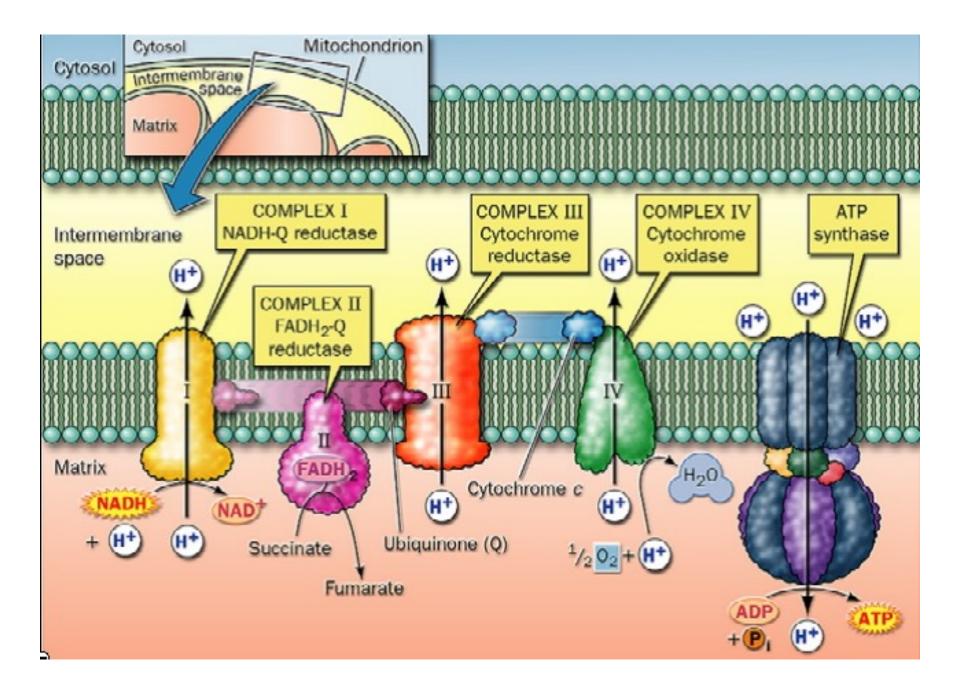
Diagram of a Mitochondrion



 Oxidative phosphorylation takes place in the inner mitochondrial membrane, in contrast with most of the reactions of the citric acid cycle and fatty acid oxidation, which take place in the matrix

The Respiratory Chain Consists of Four Complexes: Three Proton Pumps and a Physical Link to the Citric Acid Cycle



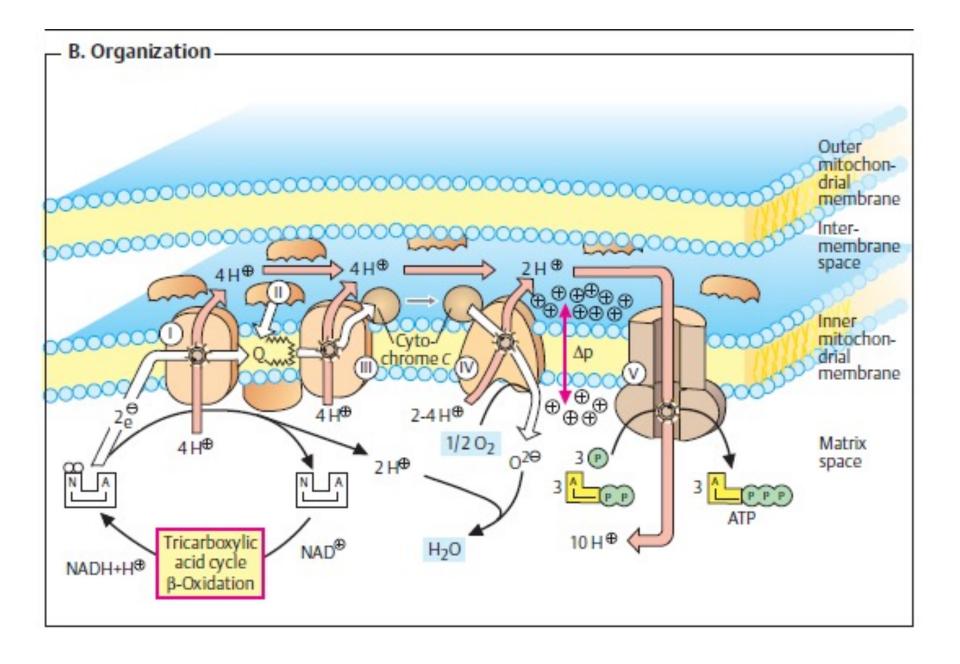


The **respiratory chain** is one of the pathways involved in *oxidative phosphorylation* It catalyzes the steps by which electrons are transported from NADH+H+ or reduced ubiquinone (QH2) tomolecular oxygen Most of the energy released is used to establish a proton gradient across the inner mitochondrial membrane, which is then ultimately used to synthesize ATP with the help of *ATP*

synthase.

Components of the respiratory chain

- The **electron transport chain** consists of:
- ✓ 3 protein complexes (complexes I, III, and IV), which are integrated into the innermitochondrial membrane, and
- ✓ 2 mobile carrier molecules— ubiquinone (coenzyme Q) and cytochrome c



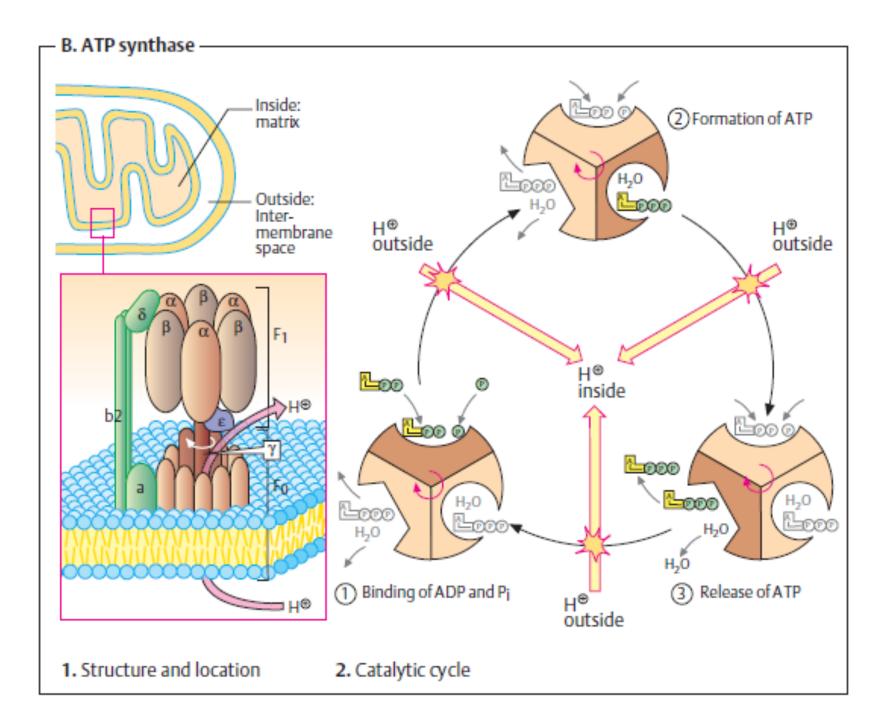


Table 18.4. ATP yield from the complete oxidation of glucose

| Reaction sequence | ATP yield per glucose molecule |
|--|--------------------------------|
| Glycolysis: Conversion of glucose into pyruvate (in the cytosol) | |
| Phosphorylation of glucose | - 1 |
| Phosphorylation of fructose 6-phosphate | - 1 |
| Dephosphorylation of 2 molecules of 1,3-BPG | + 2 |
| Dephosphorylation of 2 molecules of phosphoenolpyruvate | + 2 |
| 2 molecules of NADH are formed in the oxidation of 2 molecules of glyceraldehyde 3-phosphate | |
| Conversion of pyruvate into acetyl CoA (inside mitochondria) | |
| 2 molecules of NADH are formed | |
| Citric acid cycle (inside mitochondria) | |
| 2 molecules of guanosine triphosphate are formed from 2 molecules of succinyl CoA | + 2 |
| 6 molecules of NADH are formed in the oxidation of 2 molecules each of isocitrate, α -ketoglutarate, and malate | |
| 2 molecules of FADH ₂ are formed in the oxidation of 2 molecules of succinate | |
| Oxidative phosphorylation (inside mitochondria) | |
| 2 molecules of NADH formed in glycolysis; each yields 1.5 molecules of ATP (assuming transport of NADH by the glycerol 3-phosphate shuttle) | + 3 |
| 2 molecules of NADH formed in the oxidative decarboxylation of pyruvate; each yields 2.5 molecules of ATP | + 5 |
| 2 molecules of FADH ₂ formed in the citric acid cycle; each yields 1.5 molecules of ATP | + 3 |
| 6 molecules of NADH formed in the citric acid cycle; each yields 2.5 molecules of ATP | + 15 |
| NET YIELD PER MOLECULE OF GLUCOSE | E + 30 |

 Oxidative phosphorylation takes place in the inner mitochondrial membrane, in contrast with most of the reactions of the citric acid cycle and fatty acid oxidation, which take place in the matrix